

LETTER TO THE EDITOR

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A proposal for a new temperature-corrected formula for the oxygen content of blood

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To the Editor,

Oxygen (O₂) is essential for normal aerobic metabolism in mammals. O₂ delivery is the process of transport of oxygenated blood from the alveolar capillaries to the tissues and is dependent on two factors: arterial O₂ content (CaO₂) and cardiac output. The carriage in the blood occurs as O₂ bound to hemoglobin and O₂ dissolved in the plasma and is expressed as the sum of these components with the following equation: $CaO_2 = 1.39 \times Hb \text{ (g/dl)} \times SaO_2 + 0.0031 \times PaO_2 \text{ (mmHg)}$ (1). CaO₂ is the milliliters of O₂ per 100 mL of blood, SaO₂ is the fraction of hemoglobin (Hb) that is saturated with O₂, the O₂-combining capacity of Hb is 1.39 mL of O₂ per gram of Hb, Hb is grams of Hb per 100 mL of blood, PaO₂ is the O₂ tension, and solubility of O₂ in plasma is 0.003 mL of O₂ per 100 mL plasma for each mm Hg of PaO₂ [1, 2]. The pressure is 1 atm (101.325 kPa). However, the temperature is not clearly specified. In fact, the coefficient of 1.39 in the previous item of the equation was obtained under the standard condition of 0 °C (273 K). It is obtained as $22.4 \text{ (volume of 1 mol of gas at 273 K)} \times 4 \text{ (molecular number bound to 1 molecule of hemoglobin)} / 64,458 \text{ (molecular weight of hemoglobin)} / 1000 \text{ (correction coefficient)}$ [3, 4]. In the case of 37 °C (310 K), the coefficient should be 1.58 due to Boyle-Charles law or combined gas law $(273 + 37) / 273 \times 1.39$. On the contrary, the coefficient 0.0031 from the last item of the equation is derived from evidence on O₂ solubility at 37 °C (310 K). It is obtained as $0.0239 \text{ (Bunsen absorption coefficient } \alpha \text{ of oxygen at 37 °C)} \times 100 / 760$ [5]. These items from the same equation are derived at different temperatures. Although equation (1) is established and is described in almost all published

textbooks, the equation should be revised so that it is more clinically relevant. In lieu of equation (1), we propose a new temperature correction formula: $CaO_2 \text{ (ml/dl)} = 1.58 \times Hb \text{ (g/dl)} \times SaO_2 + 0.0031 \times PaO_2 \text{ (mmHg)}$ at 37 °C and 1 atm. Lack of experimental results is one of limitations of our proposal, and we need to confirm them in clinical settings.

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References

1. West JB. Gas transport by the blood. In: West JB, Luks AM, editors. *West's Respiratory Physiology: The Essentials* 10th Ed. Baltimore: Lippincott Williams & Wilkins; 2015. p. 75–92.

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2. Severinghaus JW. Oxyhemoglobin dissociation curve correction for temperature and pH variation in human blood. *J Appl Physiol*. 1958;12:485–6.
3. Sendroy J, Dilton RT, Van Slyke DD. Studies of gas and electrolyte equilibria in the blood : XIX. The solubility and physical state of uncombined oxygen in blood. *J Biol Chem*. 1934;105:597–632.
4. Pittman RN. Chapter 4 Oxygen transport In: Regulation of tissue oxygenation. San Rafael: Morgan & Claypool Life Sciences; 2011. <https://www.ncbi.nlm.nih.gov/books/NBK54103/> Accessed 2 August 2010.
5. Forstner H, Gnaiger E. Calculation of equilibrium oxygen concentration. In: Gnaiger E, Forstner H, editors. Polarographic oxygen sensor: aquatic and physiological applications. Berlin: Springer-Verlag; 1983. p. 321–33.

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